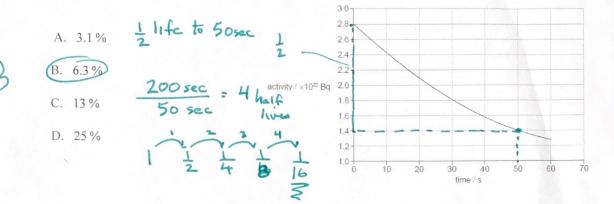
1. The graph shows the variation with time of the activity of a pure sample of a radioactive nuclide.

What percentage of the nuclide remains after 200 s?



2. Copper $\binom{64}{29}$ Cu) decays to nickel $\binom{64}{28}$ Ni). What are the particles emitted and the particle that mediates the interaction?

		Particles emitted	Mediating particle
64 Cu > 64 Ni + e° + V	A.	β ⁻ and neutrino	W ⁺
29 28 +1	В.	β ⁺ and neutrino	W-
particle w+	C.	β ⁻ and neutrino	W-
partitude CO	(D)	β* and neutrino	W ⁺

3. The following interaction is proposed between a proton and a pion.

$p^+ + \pi^- \longrightarrow K^- + \pi^+$		
r	meson	meson

The quark content of the π^- is $\bar{\rm ud}$ and the quark content of the K^- is $\bar{\rm us}$.

Three conservation rules are considered: (i.) baryon number II. Charge (iii) strangeness.

Which conservation rules are violated in this interaction?

$$\mu^- \rightarrow e^- + \nu_\mu + X$$

What is particle *X*?

Α. γ

banyon > B 0 = 0 + 0 + 0 | New "X" particle

charge > BC -1 -> -1 + 0 + 0 | nunst satisfy

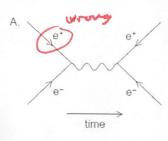
there

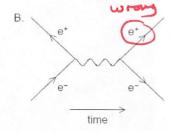
two

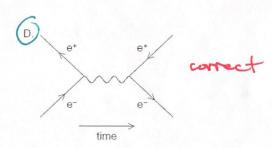
Lepton - electron

families (Lau +1 -> 0 + 1 + 0) - zero charge

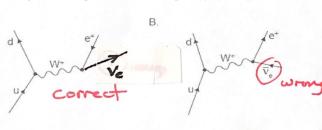
5. Which is the correct Feynman diagram for pair annihilation and pair production?

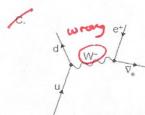


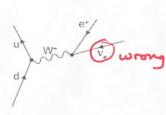




6. Which Feynman diagram shows beta-plus (β^+) decay?





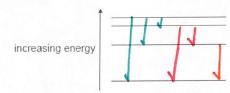


- 7. The average binding energy per nucleon of the ${}^{15}_{8}$ O nucleus is 7.5 MeV. What is the total energy required to separate the nucleons of one nucleus of ${}^{15}_{8}$ O?
 - A. 53 MeV

- B. 60 MeV
- (C.) 113 MeV
- D. 173 MeV
- 8. The energy-level diagram for an atom that has four energy states is shown.

What is the number of different wavelengths in the emission spectrum of this atom?

- A. 1
- B. 3
- (C.) (
 - D. 7



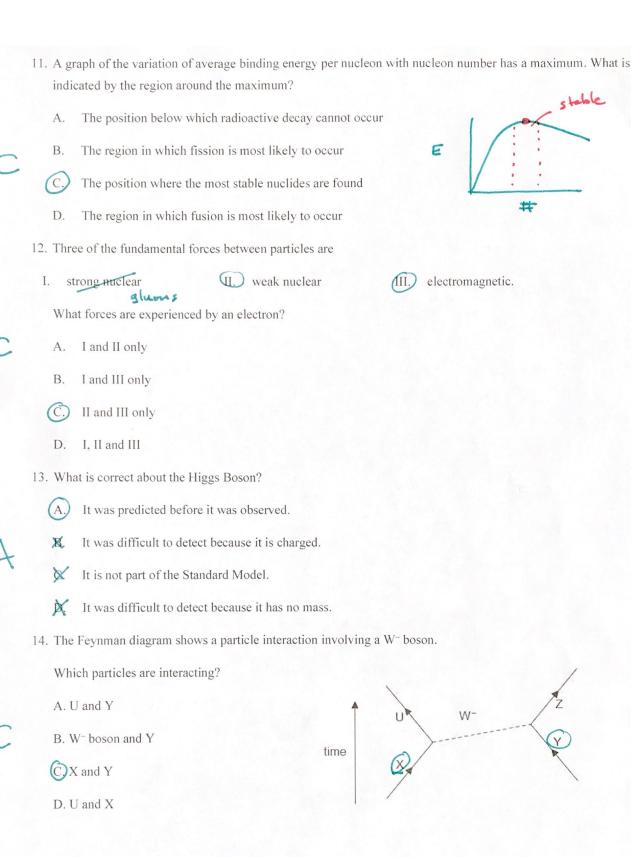
9. A detector, placed close to a radioactive source, detects an activity of 260 Bq. The average background activity at this location is 20 Bq. The radioactive nuclide has a half-life of 9 hours.

What activity is detected after 36 hours?

- A. 15 Bq
- B. 16 Bq
- C. 20 Bq
- D. 35 Bq



- 10. Element X decays through a series of alpha (α) and beta minus (β -) emissions. Which series of emissions results in an isotope of X?
 - A. 1α and 2β
 - B. 1α and 4β
 - C. 2α and 2β
 - D. 2α and 3β



15. A nucleus of phosphorus (P) decays to a nucleus of silicon (Si) with the emission of particle X and particle Y.

$$^{30}_{15}P \to {}^{30}_{14}Si + X + Y$$

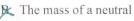
What are X and Y?

X	Y
antineutrino	positron
antineutrino	electron
neutrino	electron
neutrino	positron

16. What is the definition of the unified atomic mass unit?



 \bigcirc $\frac{1}{12}$ the mass of a neutral atom of carbon-12



The mass of a neutral atom of hydrogen-1

Q. The mass of a nucleus of hydrogen-1

17. Atomic spectra are caused when a certain particle makes transitions between energy levels. What is this particle?



A. Electron

- B. Proton
- C. Neutron
- D. Alpha particle

18. The half-life of a radioactive element is 5.0 days. A freshly-prepared sample contains 128 g of this element. After how many days will there be 16 g of this element left behind in the sample?

15 days

- A. 5.0 days
- 128 64 32 16
- B. 10 days
- C.15 days
- D. 20 days
- 19. The binding energy per nucleon of ${}^{11}_4Be$ is 6 MeV. What is the energy required to separate the nucleons of this nucleus?
 - A. 24 MeV

6 x 11 = 66 MeV

- B. 42 MeV
- C. 66 MeV
 - D. 90 MeV
- 20. A pure sample of nuclide A and a pure sample of nuclide B have the same activity at time t = 0. Nuclide A has a half-life of T, nuclide B has a half-life of 2T.

What is $\frac{\text{activity of A}}{\text{activity of B}}$ when t = 4T?

- A. 4
- 16 = -4
 - 4

C. $\frac{1}{2}$







