Conductors and Insulators

HTTPS://WWW.YOUTUBE.COM/WATCH?V=HJDJ4BGX6MW

Conductors

- ► Allow Movement of Electrons
- ▶ Good & fair Materials
- ► Metals = BEST! Ex. Aluminum, Coper, Steel
- Non-Metals are fair/semi conductors . Ex. Salt Water, Graphite



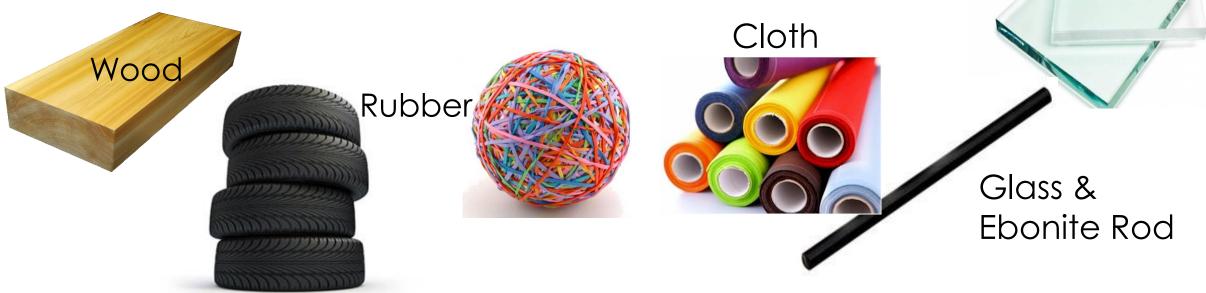


Aluminum and Coper



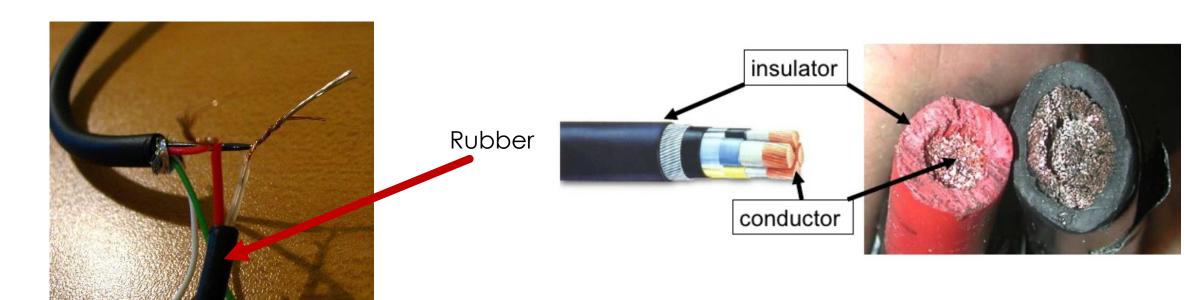
Insulators

- ▶ Reduce/ Limit the movement of electrons
- ▶ DON'T allow an electric current to pass through it
- ▶ HOLD electrons...Thus these materials HOLD any given Charge



Insulators

- ► Allow us to use electronics Safely
- ▶ Electrons are tightly bound to the atom, (Cant move very far or fast)



Thinking

- With the person beside you think of 3 things that have both conductors and insulators
- ► Previous ex. Cell phone chargers

Band Theory

- ► Electrons occupy energy levels called bands
- Conductors implies that the outer electrons of the atom are loosely bound and free to move through materials in the conduction band of an atom
- ▶ BUT the electrons need LOTS and LOTS of energy to get to this conduction band
- Metals naturally connect electrons to this conduction band, so they require less energy to get there



HOMEWORK

http://coolsciencelab.com/conductors_and_insulators.htm